

Tetrahedron Letters 41 (2000) 2609-2611

TETRAHEDRON LETTERS

## Stereocontrolled synthesis of (+)-L-noviose using a versatile sugar building block

Miwako Takeuchi, Takahiko Taniguchi and Kunio Ogasawara\*

Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980-8578, Japan

Received 11 January 2000; revised 25 January 2000; accepted 28 January 2000

## Abstract

(+)-L-Noviose, the sugar moiety of the antibiotic novobiocin, has been synthesized diastereoselectively using a man-made sugar building block. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: antibiotics; carbohydrates; enantiocontrol; neighboring group effects; stereocontrol.

The antibiotic novobiocin **1** carries a sugar moiety which is responsible for its biological activity.<sup>1</sup> The sugar moiety was found to be removed under acidic conditions to give 4-*O*-methyl-5,5-dimethyl-L-lyxose, (+)-L-noviose<sup>2</sup> **2** (Fig. 1). Although this rare sugar has been synthesized<sup>3</sup> in both racemic and enantiomeric forms, only one procedure, namely, that by Pankau and Kreiser<sup>3d,e</sup> using a chiral building block<sup>4</sup> seemed to be the most practical so far. In relation to our recent project on sugar synthesis, we chose (+)-L-noviose **2** as a target to extend the synthetic applicability of our sugar building block<sup>5,6</sup> **3** which was originally designed for the diastereocontrolled construction of eight possible aldohexoses in both enantiomeric forms.<sup>5,7</sup> We wish to report here a new synthesis of (+)-L-noviose **2** starting from our sugar building block<sup>5,6</sup> (+)-**3** which was prepared from furfural in enantiomerically pure forms by employing either a chemical<sup>5,8</sup> or an enzymatic<sup>6</sup> procedure (Scheme 1).

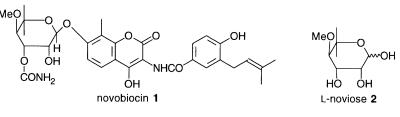
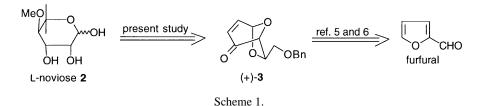


Fig. 1.

Owing to its biased structure, the building block (+)-3 having a dioxabicyclo[3.2.1] octane framework allowed diastereoselective epoxidation from the convex face on reaction with alkaline hydrogen

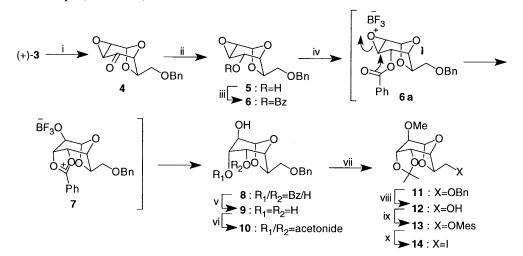
<sup>\*</sup> Corresponding author. Fax +81-22-217-6845; e-mail: konol@mail.cc.tohoku.ac.jp (K. Ogasawara)

<sup>0040-4039/00/\$ -</sup> see front matter  $\,$  © 2000 Elsevier Science Ltd. All rights reserved. PII: S0040-4039(00)00216-1



peroxide<sup>8</sup> to give the *exo*-epoxide **4** as a single product. Reduction of **4** with sodium borohydridecerium(III) chloride<sup>9</sup> also took place diastereoselectively from the convex face to give the *endo*-alcohol **5**,  $[\alpha]_D^{27} + 17.7 \ (c \ 1.1, CHCl_3)$ , as a single product. Benzoylation of **5** followed by treating the resulting benzoate **6**,  $[\alpha]_D^{29} - 18.2 \ (c \ 1.1, CHCl_3)$ , with boron trifluoride etherate<sup>10</sup> in toluene brought about regio- and diastereoselective epoxy cleavage to give the monobenzoate mixture **8**. Without separation, the mixture was subjected to alkaline methanolysis to give the single triol **9** which allowed specific ketal formation to afford the crystalline hydroxy-acetonide **10**, mp 109–110°C,  $[\alpha]_D^{29} -52.2 \ (c \ 1.1, CHCl_3)$ , as the single product. The observed diastereoselective conversion of the epoxide **6** into the single triol **9** may be readily presumed by taking account of the participation of the benzoate group which produced the benzoate mixture **8** through **6a** and **7** as shown.

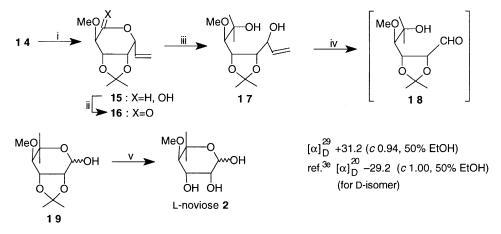
Having created three consecutive oxygen-bearing stereogenic centers, which are those required in the target molecule, the secondary alcohol **10** was first treated with methyl iodide under standard conditions to give the methyl ether **11**,  $[\alpha]_D^{29} -53.9$  (*c* 1.0, CHCl<sub>3</sub>), whose benzyl functionality was then removed under hydrogenolysis conditions to give the crystalline primary alcohol **12**, mp 83–84°C,  $[\alpha]_D^{27}$  –85.5 (*c* 1.0, CHCl<sub>3</sub>). Mesylation of **12** followed by treating the resulting mesylate **13** with lithium iodide afforded the iodide **14**,  $[\alpha]_D^{28}$  –47.8 (*c* 1.0, CHCl<sub>3</sub>), without difficulty. Overall yield of **14** from (+)-**3** was 54% in 10 steps (Scheme 2).



Scheme 2. Reagents and conditions: (i) 30%  $H_2O_2$ , 0.5N NaOH, THF, 0°C. (ii) NaBH<sub>4</sub>–CeCl<sub>3</sub>·7H<sub>2</sub>O, MeOH, 0°C (82% for two steps). (iii) BzCl, pyridine, CH<sub>2</sub>Cl<sub>2</sub> (96%). (iv) BF<sub>3</sub>·OEt<sub>2</sub>, toluene. (v) NaOMe, MeOH. (vi) Me<sub>2</sub>C(OMe)<sub>2</sub>, PPTS (cat.), toluene, reflux (84% for three steps). (vii) MeI, NaH, THF (99%). (viii) H<sub>2</sub>, 10% Pd–C, MeOH (91%). (ix) MesCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>. (x) LiI, THF, reflux (90% for two steps)

In order to attain the target molecule, the iodide **14** was treated with zinc in methanolic acetic acid to initiate reductive elimination to cleave the internal acetal functionality to give rise to the hemiacetal **15** as a mixture of epimers. Since an anhydro-sugar type internal acetal functionality requires rather severe conditions for its hydrolytic cleavage, the sugar building block **3** exerts its potential with respect

to this point. On oxidation with tetrapropylammonium perruthenate in the presence of *N*-morphorine *N*-oxide (TPAP–NMO),<sup>5,11</sup> **15** gave the  $\delta$ -lactone **16**, mp 59–60°C,  $[\alpha]_D^{29}$  +89.9 (*c* 1.1, CHCl<sub>3</sub>), as colorless needles (Scheme 3). Treatment of **16** with an excess amount of methyllithium gave the acyclic diol **17**,  $[\alpha]_D^{27}$  -51.7 (*c* 1.0, CHCl<sub>3</sub>), whose extra two-carbon moiety was oxidatively removed under Lemieux–Johnson conditions<sup>12</sup> to give the hemiacetal **19** via the transient hydroxy-aldehyde **18**. Finally, **19** was acid-hydrolyzed in the presence of Dowex 50-W<sup>13</sup> to remove the acetonide protecting group to give rise to (+)-L-noviose **2**, mp 128–129°C,  $[\alpha]_D^{29}$  +31.2 (*c* 0.94, 50% EtOH) [lit.<sup>3e</sup>: mp 128°C,  $[\alpha]_D^{20}$  –29.2 (*c* 1.00, 50% EtOH) for D-enantiomer], as colorless crystals. Overall yield of **2** from **14** was 43%.



Scheme 3. Reagents and conditions: (i) Zn, AcOH:MeOH (1:10) (97%). (ii) TPAP–NMO, 4 Å sieves,  $CH_2Cl_2$  (90%). (iii) MeLi, THF, 0°C (83%). (iv) OsO<sub>4</sub> (cat.), NaIO<sub>4</sub>, 50% aq. THF. (v) Dowex 50-W,  $H_2O$ , 70°C (59% for two steps)

In summary, we have synthesized (+)-L-noviose **2** in 23% overall yield in 15 steps with complete diastereoselection from our sugar building block (+)-**3** on the basis of its inherent convex-face selectivity and high functionality.

## References

- 1. Ferroud, D.; Collard, J.; Klich, M.; Dupuis-Hamelin, C.; Mauvais, P.; Lassaigne, P.; Bonnefoy, A.; Musicki, B. *Bioorg. Med. Chem. Lett.* **1999**, *9*, 2881.
- Shunk, C. H.; Stammer, C. H.; Kaczka, E. A.; Walton, E.; Spencer, C. F.; Wilson, A. N.; Richter, J. W.; Holly, F. W.; Folkers, K. J. Am. Chem. Soc. 1956, 78, 1770. Kaczka, E. A.; Shunk, C. H.; Richter, J. W.; Wolf, F. J.; Gasser, M. M.; Folkers, K. J. Am. Chem. Soc. 1956, 78, 4125. Hinman, J. W.; Caron, E. L.; Hoeksema, H. J. Am. Chem. Soc. 1957, 79, 5321.
- (a) Kiss, J.; Spiegelberg, H. Helv. Chim. Acta 1964, 47, 398. (b) Achmatowicz Jr., O.; Grynkiewicz, G.; Szechner, B. Tetrahedron 1976, 32, 1051. (c) Klemer, A.; Waldmann, M. Liebigs Ann. Chem. 1986, 2, 221. (d) Pankau, W. M.; Kreiser, W. Tetrahedron Lett. 1998, 39, 2089. (e) Pankau, W. M.; Kreiser, W. Helv. Chim. Acta 1998, 81, 1997.
- 4. Kreiser, W.; Wiggermann, A.; Krief, A.; Swinnen, D. Tatrahedron Lett. 1996, 37, 7119.
- 5. Takeuchi, M.; Taniguchi, T.; Ogasawara, K. Synthesis 1999, 341.
- 6. Taniguchi, T.; Takeuchi, M.; Kadota, K.; ElAzab, A. S.; Ogasawara, K. Synthesis 1999, 1325.
- 7. Takeuchi, M.; Taniguchi, T.; Ogasawara, K. Chirality in press.
- 8. Taniguchi, T.; Ohnishi, H.; Ogasawara, K. Chem. Commun. 1996, 1477.
- 9. Gemal, A. L.; Luche, J.-L. J. Am. Chem. Soc. 1981, 103, 5454.
- 10. Prystas, M.; Gustafsson, H.; Sorm, F. Coll. Czech. Chem. Commun. 1971, 36, 1487.
- 11. Ley, S. V.; Norman, J.; Griffith, W. P.; Marsden, S. P. Synthesis 1994, 639.
- 12. Pappo, R.; Allen, D. S.; Lemieux, R. U.; Johnson, W. S. J. Org. Chem. 1956, 21, 478.
- 13. Ho, P.-T. Tetrahedron Lett. 1978, 1623.